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BOOK OF ABSTRACTS



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The Reactive Species Produced by Transient Spark Discharge in Gas and Liquid Phase and Its Effect on *Escherichia coli*

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Low temperature plasma generated at atmospheric pressure in contact with water induces numerous reactive species in the gas phase, which diffuse into liquid phase and induce chemical and biological changes. Despite many studies, the exact mechanism of plasma interaction with living organisms is still not enough clear. The main objective of the study was to put into correlation the chemical changes in gas and liquid phase and bactericidal inactivation induced by the plasma generated in various gas mixtures above various water solutions and to get more information on the reactive agents responsible for the biocidal effects. We used the self-pulsing transient spark (TS) discharge driven by positive DC power supply generated in various gas mixtures of O₂/N₂ (~ 2 L/min) above the circulating water solutions (~ 14 mL/min) - non-buffered solution of monosodium phosphate (NaH₂PO₄, pH 5, 600 S/cm, W) and 2 mM phosphate buffer solution (Na₂HPO₄/KH₂PO₄, pH 7, 550 S/cm, PB) - to explore the impact of pH on chemistry. The gas phase species were analyzed by FTIR measurements. UV-VIS absorption spectroscopy was used to measure concentrations of H₂O₂ and NO₂⁻. By UV-VIS fluorescence spectroscopy we evaluated the concentration of [•]OH radical via hydroxylation of terephthalic acid. Standard colony counting method was used to evaluate the effect of the TS on Gram-negative bacteria *E. coli*. In the gas phase, in gas mixtures containing both O₂ and N₂ molecules, NO and NO₂ were detected as main products. Their concentrations increased with O₂/(O₂+N₂) ratio up to 50%, where NO and NO₂ concentrations achieved maximum of 264 ppm and 60 ppm, respectively (Fig. 1). In the case of the TS generated in pure O₂ only O₃ (~ 110 ppm) was detected. The concentrations of H₂O₂, NO₂⁻ and [•]OH radical in water solutions depended also on O₂/N₂ ratio in gas mixture. For the TS generated in pure O₂ or N₂, the NO₂⁻ concentration was low, while H₂O₂ was relatively high (Fig. 2). In the case gas mixtures contained both O₂ and N₂, the most significant pH decrease and balanced concentrations of reactive species (0.4-0.5 mM) resulted into the strongest bactericidal effect due to the formation of ONOO⁻ from reaction of H₂O₂ with NO₂⁻ [1].

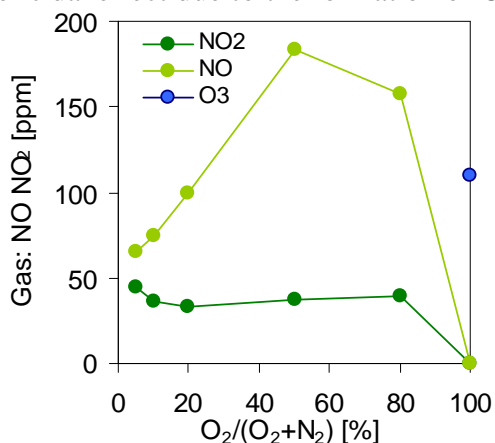


Fig. 1 Concentrations of NO and NO₂ in the gas phase generated by TS discharge.

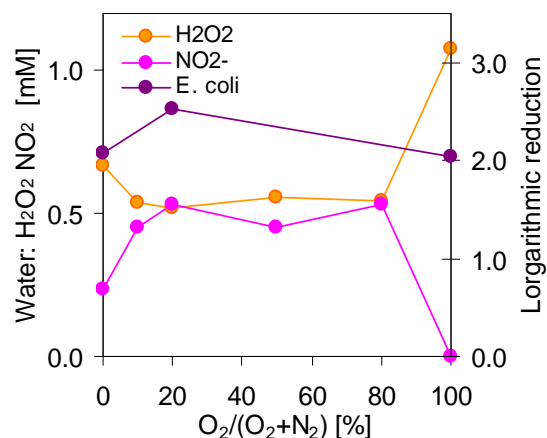


Fig. 2 Concentrations of H₂O₂, NO₂⁻ and inactivation of *E. coli* in water (W).

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References

- [1] P. Lukes, E. Dolezalova, I. Sisrová, M. Clupek, *Plasma Sources Sci. Technol.* **23**, 15 (2014).